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# Growth and Property of Stoichiometric LiNbO<sub>3</sub> Crystal<sup>†</sup>

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Near-stoichiometric LiNbO3 and Fe:LiNbO3 were grown. The ultraviolet-visible absorption spectra, infrared absorption spectra, exponent gain coefficient, diffraction efficiency and response time of these crystals were measured. It is found that the response time of Nearstoichiometric LiNbO3 and Fe:LiNbO3 was one order of magnitude less than that of congruent LiNbO<sub>3</sub>.

Keywords: LiNbO3 crystal; absorption spectrum; exponent gain coefficient

# **INTRODUCTION**

LiNbO3 crystal is electro-optic crystal with perfect electro-optic property and nonlinear optical property. It is used in many fields as a kind of important photorefractive material. That there are many intrinsic defects in the crystal are the main shortcomings of the congruent LiNbO<sub>3</sub> crystal [1]. The nearstoichiometric LiNbO<sub>3</sub> has much less intrinsic defects than the congruent LiNbO<sub>3</sub>. So the photorefractive ability of the near-stoichiometric LiNbO<sub>3</sub> is higher than that of the congruent LiNbO<sub>3</sub>.

# EXPERIMENTAL PROCEDURES AND RESULTS

#### The Growth of Crystals

Czochralski method was used to grow LiNbO3 crystal and Fe:LiNbO3 crystal. The raw materials using to grow crystals were Li<sub>2</sub>CO<sub>3</sub>, Nb<sub>2</sub>O<sub>5</sub> and Fe<sub>2</sub>O<sub>3</sub>

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with purity of 99.99%. The ratio of  $[Li_2CO_3]$  to  $[Nb_2O_5]$  was 1.44.  $[Fe_2O_3]$  was 0.03 mol%. The growth orientation was c-axis. The optimized technology conditions of crystal growth were that the temperature gradient was  $30 \sim 40^{\circ}$  C/cm, the rotating rate was  $20 \sim 30$  rpm, and the pulling rate was 0.2 mm/h. The crystals were made into samples after being cut and polished, which dimension was  $10 \times 5 \times 1$  mm<sup>3</sup>.

#### Ultraviolet-Visible Absorption Spectra of LiNbO<sub>3</sub> Crystals

The ultraviolet-visible absorption spectra of congruent  $LiNbO_3$  and nearstoichiometric  $LiNbO_3$  were measured by using CARYIE style UV-Visible Spectrophotometer. The results was shown in Fig. 1.

The absorption edges of congruent LN and near-stoichiometric crystals spectra located at 320 nm and 306 nm, respectively. The relation of the wavelength ( $\lambda$ ) of the absorption edge with the concentration ( $\chi$ ) of Li<sup>+</sup> could be described by the following equation [2]

$$\lambda = 321.9 - 1.597\chi - 5.745\chi^2 \tag{1}$$

From the last equation, the concentration of 49.6 mol% of Li<sup>+</sup> in congruent LiNbO<sub>3</sub> was obtained. Because there are much less intrinsic defects in the near-stoichiometric LiNbO<sub>3</sub> than in the congruent LiNbO<sub>3</sub>, i.e. the polarizing degree of  $O^{2-}$  was decreased, compared with the absorption edge of the congruent LiNbO<sub>3</sub>, that of near-stoichiometric LiNbO<sub>3</sub> shifted to the ultraviolet band.



Figure 1. The ultraviolet-visible absorption spectra of congruent LN and nearstoichiometric crystals.



Figure 2. The infrared absorption spectrum of the congruent LiNbO<sub>3</sub>.

#### Infrared Absorption Spectra of LiNbO<sub>3</sub> Crystals

The infrared absorption spectra of the congruent and the near-stoichiometric  $LiNbO_3$  crystals were measured by Fourier conversion infrared spectrophotometer and the results were shown in Figs. 2 and 3, respectively.

For LiNbO<sub>3</sub> crystals, the position of the OH<sup>-</sup> absorption peak was affected by OH<sub>Li</sub> (H<sup>+</sup> occupied the Li<sup>+</sup> site). The main form of the intrinsic defect was Li vacancy (V<sub>Li+</sub>) and antisite Nb (Nb<sub>Li</sub><sup>4+</sup>, Nb<sup>5+</sup> occupied the Li<sup>+</sup> site). The defect structure group that one Nb<sub>Li</sub><sup>4+</sup> was connected with its nearest four Li vacancies made the charge balance in the crystal. The stretch vibrations of the defect structure group and OH<sup>-</sup> were corresponding to the peak that was located at 3483 cm<sup>-1</sup>.

#### **Exponential Gain Coefficient Measurements** [3]

The exponential gain coefficient of the photorefractive crystal is corresponding to the energy transition from the pump light to the signal light during the information storage process. The larger the exponential gain coefficient



Figure 3. The infrared absorption spectrum of the near-stoichiometric LiNbO.

 TABLE I The exponential gain coefficients of crystals

Crystal	20	$\Gamma_{max}$ (cm <sup>-1</sup> )	
Near-stoichiometric Fe:LiNbO3	32°	43	
Near-stoichiometric LiNbO3	23°	24	
Congruent LiNbO <sub>3</sub>	10°	9.5	

is, the more multiple the signal light is amplified. The exponential gain coefficient  $\Gamma$  was defined as

$$\Gamma = \frac{1}{\delta} \ln \frac{I_1' I_2}{I_1 I_2'} \approx \frac{1}{\delta} \ln \frac{I_1'}{I_1}$$
(2)

where  $I'_1$  and  $I'_2$  ( $I_1$  and  $I_2$ ) are the transmitted intensities of signal beam and pumping beam with (without) coupling, respectively. The exponential gain coefficients of the reduced samples of the near-stoichiometric Fe:LiNbO<sub>3</sub>, near-stoichiometric LiNbO<sub>3</sub> and congruent LiNbO<sub>3</sub> were measured with the two-wave coupling method. The results were shown in Table I.

#### **Diffraction Efficiency and Response Time Measurements**

The diffraction efficiency of crystals was measured by two-wave coupling method.  $Ar^+$  laser with wavelength of 514.5 nm was used, which intensity was 1.34 W/cm<sup>2</sup>. The measurement results were shown in Table II. The response time is defined as the time period that is from the time of the incident light beginning to shot on the crystal to the time when the diffraction efficiency reaches its maximum. The measurement results were shown in Table II.

**TABLE II** The diffraction efficiency and response time of crystals

Crystal	2 <del>0</del>	$\eta_{\max}$	t (s)
Near-stoichiometric Fe:LiNbO3	20.5°	70%	5.6
Near-stoichiometric LiNbO3	33°	42%	1.5
Congruent LiNbO3	9.5°	81%	32



Figure 4. The erasure curve of the near-stoichiometric LiNbO<sub>3</sub>.

#### Judgment of Styles of Light Excited Carrier [4]

When the electron was the main style of the light excited carrier in the erasure process, the erasure time of the signal light  $I_1$  is shorter than that of the pump light. When the cavity was the main style of the light excited carrier in the erasure process, the erasure time of the signal light  $I_1$  is longer than that of the pump light  $I_2$ . The erasure curves of near-stoichiometric Li were shown in Fig. 4. From Fig. 4, it could be seen that the erasure speed of the near-stoichiometric LiNbO<sub>3</sub> was faster than that of the congruent LiNbO<sub>3</sub>. So the electron was the main style of the light excited carrier in the erasure process in near-stoichiometric LiNbO<sub>3</sub> crystal.

# CONCLUSIONS

There are many intrinsic defects in the congruent LiNbO<sub>3</sub> crystal, the main styles of which are  $V_{Li+}$  and  $Nb_{Li}^{4+}$ . In the near-stoichiometric LiNbO<sub>3</sub>, the defect number of  $V_{Li+}$  and  $Nb_{Li}^{4+}$  are significantly reduced. Compared with the congruent LiNbO<sub>3</sub>, the ultraviolet-visible absorption edge of the near-stoichiometric LiNbO<sub>3</sub> shifted to ultraviolet band. The infrared absorption peak of the near-stoichiometric LiNbO<sub>3</sub> is shifted from 3483 cm<sup>-1</sup> to 3466 cm<sup>-1</sup>. The exponential gain coefficient of the near-stoichiometric LiNbO<sub>3</sub> was 24 cm<sup>-1</sup>, which is 2.5 times higher than that of the congruent LiNbO<sub>3</sub> achieved 42%. In Comparison with the congruent LiNbO<sub>3</sub>, the response time of near-stoichiometric LiNbO<sub>3</sub> became shorter. The measurement results of the near-stoichiometric Fe:LiNbO<sub>3</sub> was 43 cm<sup>-1</sup>,  $\eta_{max}$  was 70% and *t* was 5.6 seconds.

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